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uranium atoms. radiation. the  
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protons and, with the exception of  ${}^1_1\text{H}$ , neutrons.

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journey into the center of the atom begins with a brief review. You learned in Chapter 3 that the protons and neutrons in each atom are found in a tiny, central nucleus that measures about  $1/100,000$  the diameter of the atom itself. You also learned

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## Chapter 25 - Nuclear Chemistry

Radioactivity • Radioactivity is the process by which nuclei emit particles and rays as they break down. • The name of the penetrating rays emitted by a radioactive source is called radiation. • A radioactive isotope is an unstable atom which breaks down on its own, releasing energy and/or

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Chapter 10 Nuclear Chemistry Section  
10.1 Radioactivity (pages 292–297) This  
section discusses the different types of  
nuclear radiation and how they affect  
matter.

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## **Section 10.1 Radioactivity ...**

806 Chapter 25 Nuclear Chemistry

Figure 25-2 Both Pierre and Marie Curie played important roles in founding the field of nuclear chemistry Marie Curie went on to show that unlike chemical reactions, radioactivity is not affected by changes in physical conditions such as temperature and pressure She is the

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