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Print Page ASSESSMENT QUESTIONS: Questions & Answers \pm 1. Which set of diagrams shows a balanced chemical equation? A. Diagram A B. Diagram B C. Diagram C D. Diagram D Explanation: In answer C, each side contains 4 atoms of hydrogen (H), 2 atoms of oxygen (O), and 3 atoms of sulfur (S). This is balanced because there is the same number of each type of atom on each side of the equation.

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Correct Answer: C. Single replacement When hydrogen peroxide (H_2O_2) is added to potassium iodide (KI) solution, the hydrogen peroxide decomposes into water (H_2O) and oxygen (O_2). The

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chemical equation for this decomposition reaction is:

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Practice balancing chemical equations by changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.

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Gizmo Answer Key Chemical Equations

To set up an equation in the Chemical Equations Gizmo, type the chemical formulas into the text boxes of the Gizmo. First, type in "H₂+O₂" in the Reactants box and "H₂O" in the Products box.

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In the Chemical Changes Gizmo, you will look for evidence of chemical changes by looking at changes you can see, touch, or smell. To begin, check that Reactant 1 is Sodium and Reactant 2 is Water.

Student Exploration- Chemical Changes (ANSWER KEY) by

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If you also get perplexed in balancing chemical equations, follow
the tips for correct balancing chemical equations worksheet
answers. Tip # 1: When you are trying to balance the chemical
equations, you should remember that you can only change the
value of coefficient in front of the element or compound, and not
the subscript.

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