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- 2 - Section A Answer all questions. Write your answers in the boxes provided. 1. Ethanedioic acid is a diprotic acid. A student determined the value of x in the formula of hydrated ethanedioic acid,  $\text{HOOC-COOH}\cdot x\text{H}_2\text{O}$ , by titrating a known mass of the acid with a 0.100 mol dm<sup>-3</sup> solution of NaOH (aq). 0.795 g of ethanedioic acid was dissolved in distilled water and made up to a total ...

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2 1. d slow rate of reaction which gradually increases as H ions are produced «to catalyse the reaction» OR reaction is autocatalytic M1 should mention "rate of reaction". 2 2. a electrostatic attraction between «a lattice of» metal/positive ions/cations AND «a sea of» delocalized electrons Accept "mobile electrons".

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- 3 - Turn over (Question 1 continued) (ii) 3A student produced these results with  $[\text{H}^+] = 0.15 \text{ mol dm}^{-3}$ . Propanone and acid were in excess and iodine was the limiting reagent.

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